2005 Vol. 7, No. 10 2023–2025

Xenibellols A and B, New Diterpenoids from the Formosan Soft Coral *Xenia umbellata*

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Received March 10, 2005

ABSTRACT

Xenibellols A (1) and B (2), possessing an unprecedented diterpenoid skeleton, were isolated from the soft coral *Xenia umbellata*. The structures of xenibellols A and B were established by extensive analysis of spectroscopic data.

Soft corals of the genus Xenia are rich in diterpenoids.¹ As part of our search for novel bioactive substances from marine and terrestrial organisms,^{2–4} the soft coral *Xenia umbellata* Lamarck was studied because CH₂Cl₂ extracts showed significant cytotoxicity to A549 (human lung adenocarcinoma), HT-29 (human colon adenocarcinoma), and P-388 (mouse lymphocytic leukemia) cell cultures as determined by standard procedures.^{5,6} Bioassay-guided fractionation resulted in the isolation of two novel cytotoxic diterpenoids (novel carbon skeleton), xenibellols A (1) and B (2).

The soft coral *X. umbellata* was collected at Green Island, Taiwan, in September, 2004, at a depth of 3–4 m. The bodies of the soft coral were freeze-dried to give 1.20 kg of a solid,

which was extracted with acetone— CH_2Cl_2 (5.0 L × 3). After removal of solvent in vacuo, the residue (66 g) was chromatographed over silica gel 60 using *n*-hexane and *n*-hexane—EtOAc mixtures of increasing polarity. Elution by *n*-hexane—EtOAc (1:9) afforded a fraction containing compounds **1** and **2**. Compounds **1** (12 mg, 0.018%) and **2** (4 mg, 0.006%) were further purified by HPLC (LiChrosorb RP-18, 7 μ m, 25 × 250 mm), eluting with MeOH—H₂O (65:35).

Compound **1** was isolated as a colorless oil, $[\alpha]^{25}_D + 18^\circ$ (c 0.1, CHCl₃). The IR spectrum of **1** exhibited absorptions due to hydroxyl (3420 cm⁻¹), lactone carbonyl (1732 cm⁻¹), and conjugated diene (1640 cm⁻¹) groups. The presence of the conjugated diene was also confirmed by the UV spectrum [λ_{max} 228 nm (ϵ 13 100)]. HRESIMS suggested a molecular formula of $C_{20}H_{28}O_4$ ([M + Na]⁺, m/z 355.1886, Δ +0.0002 mmu).

The structure of 1 was completely assigned by a combination of one- and two-dimensional NMR methods. The carbon

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Table 1. ¹H NMR (300 MHz) and ¹³C NMR (75 MHz) Data^a of 1 and 2

1			2		
position	$\delta_{ m H},{ m mult}(J{ m inhertz})$	$\delta_{\mathrm{C}}\left(\mathrm{mult}\right)$	position	δ_{H} , mult (J in hertz)	$\delta_{C}\left(mult\right)$
1		173.0 (qC)	1	9.45 s	203.3 (CH)
3	4.45 d (12.0), 4.98 d (12.0)	$73.0 (\mathrm{CH_2})$	3	$9.37 \mathrm{\ s}$	195.2 (CH)
4		133.1 (qC)	4		141.5 (qC)
4a	2.95 m	35.1 (CH)	4a	3.11 m	31.0 (CH)
5	2.21 m, 1.64 m	$28.5 (CH_2)$	5	1.52 m	$25.0 (CH_2)$
6	1.56 m	$29.5 (CH_2)$	6	1.53 m	$28.7 (CH_2)$
7		48.6 (qC)	7		48.6 (qC)
8	$3.84 \mathrm{\ s}$	82.9 (CH)	8	$3.85 \mathrm{\ s}$	82.5 (CH)
9	1.88 m	$28.0 (CH_2)$	9	1.37 m, 1.82 m	$28.4 (CH_2)$
10	1.95 m	$26.8 (CH_2)$	10	1.47 m, 2.10 m	$27.3 (CH_2)$
11		47.6 (qC)	11		48.2 (qC)
11a	2.61 d (12.0)	42.2 (CH)	11a	3.45 d (12.0)	49.7 (CH)
12	6.13 d (11.4)	129.6 (CH)	12	6.88 m	152.1 (CH)
13	6.43 dd (15.0, 11.4)	121.0 (CH)	13	6.91 m	121.2 (CH)
14	5.90 d (15.0)	145.0 (CH)	14	6.39 dq (14.1, 3.6)	153.1 (CH)
15		71.0 (qC)	15		71.3 (qC)
16	$1.37 \mathrm{\ s}$	$30.1 (CH_3)$	16	$1.42 \mathrm{\ s}$	29.7 (CH ₃)
17	$1.37 \mathrm{\ s}$	$30.1 (CH_3)$	17	$1.44 \mathrm{\ s}$	29.9 (CH ₃)
18	$1.19 \mathrm{\ s}$	$15.8 (CH_3)$	18	$1.27 \mathrm{\ s}$	15.1 (CH ₃)
19	3.50 dd (7.2, 2.7), 4.06 d (7.2)	$74.9 (\mathrm{CH_2})$	19	3.71 m	$75.4~(CH_2)$

^a Spectra were recorded in CDCl₃ at 25 °C. Chemical shift values are given in parts per million relative to TMS. ¹³C NMR multiplicities were obtained from a DEPT-135 experiment.

resonances at $\delta_{\rm C}$ 133.1 (qC), 129.6 (CH), 121.0 (CH), and 145.0 (CH), in the $^{13}{\rm C}$ NMR and DEPT spectra showed the presence of a diene (Table 1). Furthermore, the presence of four sp³ oxygenated carbons was inferred from the carbon signals at $\delta_{\rm C}$ 71.0 (qC), 73.0 (CH₂), 74.9 (CH₂), and 82.9 (CH). Four methylene groups were deduced from the DEPT signals at $\delta_{\rm C}$ 26.8, 28.0, 28.5, and 29.5, two methine signals at $\delta_{\rm C}$ 35.1 and 42.2, two quarternary carbon signals at $\delta_{\rm C}$ 47.6 and 48.6, a lactone carbonyl at δ 173.0 (qC), and, finally, two methyl signals at $\delta_{\rm C}$ 30.1 (CH₃).

The 1 H NMR spectrum confirmed the presence of a conjugated diene by the fact that three signals were observed at $\delta_{\rm H}$ 5.90 (d, J=15.0 Hz), 6.13 (d, J=11.4 Hz), and 6.43 (dd, J=15.0, 11.4 Hz). In addition, two oxygenated methylenes were observed at $\delta_{\rm H}$ 4.98 (d, J=12.0 Hz), 4.45 (d, J=12.0 Hz), and $\delta_{\rm H}$ 3.50 (d, J=7.2 Hz), 4.06 (d, J=7.2 Hz). One intense singlet signal is also observed at $\delta_{\rm H}$ 1.37 (s, 6H), and this correspond to two methyl groups. In this manner, the seven degrees of unsaturation present in 1 were established.

The combined use of ${}^{1}H^{-1}H$ COSY and HMQC on 1 allowed us to distinguish three spin systems (see $\mathbf{a}-\mathbf{c}$ in Figure 1) and two methyl groups linked to an oxygenated quaternary carbon. An HMBC experiment was used to assemble the skeletal fragments through quaternary carbons and heteroatoms. Thus, these substructures were connected through HMBC correlations between the protons H_2 -3 (δ_H 4.45 and 4.98) and the carbons C-1 (δ_C 173.0), C-4 (δ_C 133.1), C-12 (δ_C 129.6), and C-4a (δ_C 35.1), between the protons H_2 -19 (δ_H 3.50 and 4.06) and the carbons C-10 (δ_C 26.8), C-7 (δ_C 48.6), C-11 (δ_C 47.6), and C-8 (δ_C 82.9), between the methylene protons H_2 -9 (δ_H 1.88) and carbons

C-11 and C-7, between the protons H-11a and the carbon C-1 and C-4, between the methyl protons Me-18 ($\delta_{\rm H}$ 1.19)

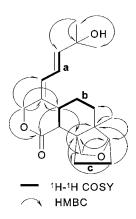


Figure 1. Key ${}^{1}H-{}^{1}H$ COSY and HMBC correlations of 1.

and carbons C-6 ($\delta_{\rm C}$ 29.5), C-7 ($\delta_{\rm C}$ 48.6), and C-8 ($\delta_{\rm C}$ 82.9), and between the methine proton H-4a ($\delta_{\rm H}$ 2.95) and carbon C-4/C-11. These relationships are represented in Figure 1.

All these data allowed us to identify compound 1 as a new diterpenoid with a novel skeleton. With the gross structure of 1 in hand, the relative stereochemistry of compound 1 was deduced from NOESY correlations (Figure 2) and by comparison of its spectroscopic data to those of xenia diterpenes.^{7–9} E geometry was assigned to the $\Delta^{4,12}$ double bond on the basis of the observation of a NOESY correlation between H-3 α and H-12 and between H-13 and

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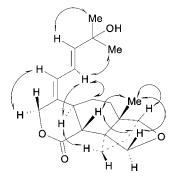


Figure 2. Key NOESY correlations of 1.

H-4a/H-5. The *E* geometry of the Δ^{13} double bond was established by the large coupling constant observed between H-13 and H-14 ($J=15.0~{\rm Hz}$). The coupling constant ($J=12.0~{\rm Hz}$) between H-4a and H-11a suggested a trans ring junction, which implied that H-4a was α-oriented.⁷⁻⁹ NOE correlations from Me-18 to H-8/H-11a/H-19β and from H-4a to H-10 were observed. This suggests that H-11a, H-19β, H-8, and Me-18 are on the β face of the molecule, while H-4a, H₂-10, and H₂-9 are on the opposite (α) face of the molecule.

Compound **2** was isolated as a colorless oil, $[\alpha]^{25}_D + 15^\circ$ (c 0.1, CHCl₃). The IR spectrum of **2** exhibited absorptions due to hydroxyl (3480 cm⁻¹) and $\alpha, \beta, \gamma, \delta$ -unsaturated aldehyde (1710, 1660 cm⁻¹) groups. The presence of the $\alpha, \beta, \gamma, \delta$ -unsaturated aldehyde was also confirmed by the UV spectrum [λ_{max} 233 nm (ϵ 11 200)]. HRESIMS suggested a molecular formula of $C_{20}H_{28}O_4$ ([M + Na]⁺, m/z 355.1887, +0.0003 mmu).

The NMR data of **2** were analogous to those of **1** except that the δ -lactone was replaced by two aldehyde groups. The carbon resonances at $\delta_{\rm C}$ 195.2 (qC), 141.5 (qC), 152.1 (CH), 121.2 (CH), and 153.1 (CH) in the $^{13}{\rm C}$ NMR and DEPT spectra of **2** showed the presence of an $\alpha,\beta,\gamma,\delta$ -unsaturated aldehyde, and the $^{1}{\rm H}$ NMR spectrum of **2** confirmed the presence of an $\alpha,\beta,\gamma,\delta$ -unsaturated aldehyde by the fact that signals were observed at $\delta_{\rm H}$ 6.91, 6.88, 6.39, and 9.37. HMBC correlations between the protons H-3 and the carbons C-4/C-12/C-4a confirmed the location the $\alpha,\beta,\gamma,\delta$ -unsaturated aldehyde (Figure 3). COSY correlation between H-1 and H-11a and HMBC correlation between H-1 and C-11a showed the presence of the other aldehyde at C-11a (Figure 3).

The relative stereochemistry of compound **2** was deduced from NOESY correlations (Figure 4) and by comparison of its spectroscopic data to those of xenia diterpenes.^{7–10} E geometry was assigned to the $\Delta^{4,12}$ and $\Delta^{13,14}$ double bonds

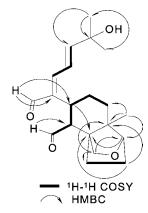


Figure 3. Key ¹H-¹H COSY and HMBC correlations of 2.

by comparing the ¹H and ¹³C NMR data with those of xenia diterpenes containing similar side chains. ^{7–10} The NOE

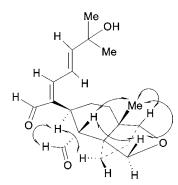


Figure 4. Key NOESY correlations of 2.

correlations from Me-18 to H-8/H-11a/H-19 β and NOE correlations from H-4a to H-1/H-10 were observed. This suggests that H-11a, H-19 β , H-8, and Me-18 are on the β face of the molecule, while H-4a, H₂-10, and H₂-9 are on the opposite (α) face of the molecule.

Xenibellols A (1) and B (2) exhibited cytotoxicity against P-388 cell with ED₅₀ of 3.6 and 2.8 μ g/mL, respectively. Biogenetically, xenibellol A (1) may be a double cyclization ([2 + 2 + 2] reaction) product of the hydroxy analogue of xeniolide-G.⁹ The dialdehyde of xenibellol B (2) was suggested as the precursor of the lactone ring of xenibellol A (1).¹⁰

Acknowledgment. We thank J. M. Pezzuto, Department of Medicinal Chemistry and Pharmacognosy, University of Illinois, for providing the P-388 cell line. This work was supported by grants from the National Science Council of Taiwan awarded to C.-Y.D.

Supporting Information Available: ¹H NMR and ¹³C NMR spectra of **1** and **2**. This material is available free of charge via the Internet at http://pubs.acs.org.

OL0505205

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